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This video is designed to help you complete the Lewis Structures Chemquest. Remember to write down the definitions and take down the notes. Circle and write down any questions you have so you do ...

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### **ChemQuest 23 : Drawing Covalent Compounds**

Information: Steps for Drawing Lewis Structures for Covalent Compounds Study the two examples in the table of how to write structures for CO<sub>3</sub><sup>2-</sup> and NH<sub>3</sub>. Make sure you understand each of the five steps. CO<sub>3</sub><sup>2-</sup>-NH<sub>3</sub> Step #1: Add up the number of valence electrons that should be included in the Lewis Structure.  $4 + 3(6) + 2 = 24$  (carbon has four ...

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Therefore, the charge on each X is 1. Here's why: Remember Answer: + that S has a negative 2 charge. There are two "X" ions to balance out the negative 2 charge of S.

### **Unit 7: Lewis Structures Chemquest**

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### **24 ChemQuest 8 Name: - Yumpu**

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24 ChemQuest 29 Monc LcW s Structures tr l Name Date: Hour: Introduction Questions Draw the Lewis structures for the following molecules. Part c is done for you. a) H<sub>2</sub>O (oxygen is central) b) SO<sub>2</sub> c) N<sub>2</sub> It t-t (a [Jrt @ N=N @ @ f\ @ 2 In question 1 above you should see three different kinds of bonds as best you can. a) single bond: one es ...

### Chemquest 24 More Lewis Structures

24 ChemQuest 8 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Structure of the Atom Note the following symbols: (they are not to scale) = proton (positive charge) = electron (negative charge) = neutron (no charge) The following three diagrams are hydrogen atoms: 1 1H 2 1H 3 1H The following three diagrams are carbon atoms: Notice the type of notation used for atoms: 12 13 14 6C 6 C 6C (6 ...

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Note: the only difference in the resonance structures is the position of the double bonds. Critical Thinking Questions 9. a) The nitrate ion, NO<sub>3</sub><sup>-</sup>, is best represented by three resonance structures. Draw the three structures below. Note: the only difference between the three structures you draw will be the position of the double bond.

### Atomic Structure - Part 2 - CHM 1032

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Information: Steps for Drawing Lewis Structures for Covalent Compounds Study the two examples in the table of how to write structures for CO<sub>3</sub><sup>2-</sup> and NH<sub>3</sub>. Make sure you understand each of the five steps. CO<sub>3</sub><sup>2-</sup>-NH<sub>3</sub> Step #1: Add up the number of valence electrons that should be included in the Lewis Structure.  $4 + 3(6) + 2 = 24$  (carbon has four ...

### **Worksheet: More Lewis Structures Name**

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### **Bonding | mretter**

Atomic Structure - Part 2. ChemQuest #24: Ionic Bonding ChemQuest #25: Advanced Ionic Bonding ChemQuest #26: Transition Metals Read: ... ChemQuest #28: Lewis Structures ChemQuest #29: More Lewis Structures ChemQuest #31: Molecular Geometry ChemQuest #32: Electronegativity Read:

### **Information : Drawing Covalent Compounds**

Worksheet: More Lewis Structures Name\_\_\_\_\_ CHEMISTRY: A Study of Matter © 2004, GPB 5.14 I. Draw Lewis Structures for the following molecules.